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A new microwave ceramic with low-permittivity for LTCC applications

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Abstract

A new low-temperature microwave dielectric ceramic with low-permittivity was found and investigated in the CaO–MgO–SiO₂–TiO₂ system. Phase constitutes and dielectric properties of $(Ca_{1-x}Mg_x)SiO_3(x=0.1-0.5)$ ceramics were studied with different *x* contents, and CaTiO₃ demonstrated an effective compensation in $(Ca_{1-x},Mg_x)SiO_3 \tau_f$ value. The 0.88CaMgSi₂O₆–0.12CaTiO₃ compositions show a relatively low-permittivity (∼9.42), high $Q \times f$ values up to 52,800 GHz, and near-zero temperature coefficients (5.6 ppm/ \degree C), which were obtained via sintering at 1300 \degree C. The addition of ≤2.0 wt% Li₂CO₃-V₂O₅ was very effective in lowering the sintering temperature (*T_s*), and dense ceramics could be obtained at T_s ≤ 900 °C. The addition of Li₂CO₃–V₂O₅ does not induce apparent degradation in the microwave properties. The specimens with 1 wt% Li₂CO₃–V₂O₅ sintered at 880 °C for 2 h shows excellent dielectric properties: $\varepsilon_r = 9.23$, $Q \times f = 46,200$ GHz, and $\tau_f = 1.3$ ppm/°C. The low-temperature sintering ceramics powders were suitable for the tape casting process. Also, the material is compatible with Ag electrodes, and therefore, is suitable for LTCC application.

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1. Introduction

The rapid growth of the telecommunication and satellite broadcasting industry has created a high demand for microwave ceramic components. Many materials have been developed and modified, according to the needs of the specific applications. Recently, considerable attention has been paid to the development of the low-temperature co-fired ceramics (LTCC) for the benefits offered for the fabrication of miniature multilayer devices. Several LTCC microwave dielectric ceramics systems such as $BiNbO₄$, $MTiO₃$ (M = Mg, Zn, Ca), BaO–TiO2, ZnNb2O6, Ca[(Li1/3Nb2/3) ¹−*x*Ti*x*]O3−δ, $BaO-R_2O_3-TiO_2$ (R = Nd, Sm), $Li_{1+x-y}Nb_{1-x-3y}Ti_{x+4y}O_3$, etc., have been developed over the years.^{[1–12](#page-4-0)} However, most of the reported LTCC microwave dielectric ceramics show relatively higher dielectric constant (>18), usually applied in low frequency ranges such as 1.89 and 2.45 GHz. Comparatively, LTCC microwave dielectric ceramics with low-permittivity,

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which are required for multilayer components particularly antennas and baluns working at a higher frequency such as 5.8 GHz, have not been extensively studied. Umemura et al.[13,14](#page-4-0) reported the dielectric properties of solid solutions in the system $(Mg_{3-x}Co_x)(VO_4)_2$. $MgCo_2(VO_4)_2$ ceramics sintered 950 °C for 5 h has $\varepsilon_r = 9.5$, $Q \times f = 78,906 \text{ GHz}$, and $\tau_f = -94.5$ ppm/ \degree C. Bian et al.^{[15](#page-4-0)} conducted a detailed investigation on the dielectric properties of AMP_2O_7 (A = Ca, Sr; $M = Zn$, Cu). Sr ZnP_2O_7 has $\varepsilon_r = 7.06$, $Q \times f = 52,781$ GHz, and τ_f = −70 ppm/ \degree C. Although the two LTCC materials have relatively low-permittivity and high $Q \times f$ values, the high coefficient of temperature resonant frequency places a constraint on their manufacture for use in practical applications. Jan-tunen et al.^{[16](#page-4-0)} reported that MgTiO₃–CaTiO₃ ceramic with a large number of ZnO–B₂O₃–SiO₂ glasses sintered at 875 °C has $\varepsilon_r = 10.6$, and $Q \times f = 7000$ GHz. Recently, Kim et al.¹⁷ acquired a new LTCC ceramics of CaWO₄ (ε _r = 8.7, $Q \times f$ = 70,220 GHz, and $\tau_f = -15$ ppm^oC). However, the material price is too high. Moreover, many kinds of LTCC materials with lowpermittivity have been developed in the form of glass + ceramics (multiphase ceramics)^{[18–21](#page-4-0)} and glass–ceramics (crystalliz-able glass).^{[22–25](#page-4-0)} Typical glass + ceramics systems include the

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borosilicate glass + alumina system developed by Fujitsu, 18 the lead borosilicate glass + alumina system by Du Pont,^{[19](#page-4-0)} and the $B_2O_3-K_2O-SiO_2-CaO-SrO-BaO$ glass + alumina system by Motorola.^{20,21} However, for the glass–ceramics system, a crystallizable glass was used, e.g., crystallizable cordierite by IBM^{22} IBM^{22} IBM^{22} and crystallizable CaO–B₂O₃–SiO₂ by Ferro.^{[24](#page-4-0)} Several these LTCC materials^{[16,21](#page-4-0)} such as DuPont 951 (ε_r = 7.8, tan δ = 0.0015 at 1 kHz), Ferro A6 (ε _r = 5.9, tan δ = 0.002 at 10 MHz), Heraeus CT 700 (ε_r = 7.0, tan δ = 0.002 at 1 kHz), Motorola T2000 $(\varepsilon_r = 9.1$, tan $\delta = 1 \times 10^{-3}$ at 2 GHz), etc., have been used for ceramics packaging and substrate. However, these LTCC materials cannot meet the requirement of microwave dielectric components due to the low $Q \times f$ values.

In this paper, we report a new LTCC microwave dielectric ceramic system CaO–MgO–SiO₂–TiO₂ with the high $Q \times f$ value, low-cost material, and the compatibility of the ceramic sheet with Ag electrodes. Phase constitutes and dielectric properties of $(Ca_{1-x},Mg_x)SiO_3$ ($x=0.1-0.5$) ceramics were studied with different x contents, and $CaTiO₃$ demonstrated an effective compensation in $(Ca_{1-x},Mg_x)SiO_3 \tau_f$ value. In addition, we utilized a very small quantity of $Li₂CO₃–V₂O₅$ combined additives in an attempt to prepare ceramics with even lower sintering temperatures down to 900 \degree C that could have potential applications as co-fired circuit components. The properties of ceramics slurry, green tape, and the chemical compatibility of silver electrodes, and the LTCC materials have also been investigated.

2. Experimental procedure

Samples of (Ca_{1-x},Mg_x) SiO₃ ($x=0.1, 0.2, 0.3, 0.4, 0.5$), and $CaTiO₃$ were individually synthesized by conventional solid-state methods from reagent-grade oxide powders (purity higher than 99%): MgO, CaCO₃, TiO₂, and SiO₂. The starting materials were mixed according to the desired stoichiometry: (Ca1−*x*,Mg*x*)SiO3, and CaTiO3, and ground in ethanol for 16 h in a ball mill with zirconia balls. Both mixtures were dried and calcined at $1100\degree$ C for 4 h in air with subsequent ball milling in ethanol with zirconia balls. After drying and sieving, the calcined (Ca_{1−*x*},Mg_{*x*})SiO₃ powders were uniaxially pressed under a pressure of 100 MPa into disks measuring 18 mm in diameter and 9 mm thickness. The disks covered with crucibles were sintered at 1300 ℃ for 3 h. The calcined reagents were individually re-milled for 16h according to the desired composition: $(1 - y)$ CaMgSi₂O₆-*y*CaTiO₃ and 0.88CaMgSi₂O₆-0.12CaTiO₃ with 0.5-2 wt% $0.88CaMgSi₂O₆ - 0.12CaTiO₃$ $Li_2CO_3-V_2O_5$. Following the same drying and forming procedures, the $(1 - y)$ CaMgSi₂O₆-*y*CaTiO₃ and 0.88CaMgSi₂O₆-0.12CaTiO₃ containing Li₂CO₃-V₂O₅ $0.88CaMgSi₂O₆ - 0.12CaTiO₃$ pellets were sintered at 1300 ◦C for 3 h and 860–960 ◦C for 2 h, respectively.

The crystalline phases were analyzed by X-ray powder diffraction analysis (XRD) using Cu K α radiation of 2 θ from 10◦ to 80◦. The microstructure observation of the samples was performed by scanning electron microscopy (SEM). The bulk densities of the sintered pellets were measured by the Archimedes method. Microwave dielectric constant ε_r and quality factor values $Q \times f$ at microwave frequencies were measured by the Hakki–Coleman dielectric resonator method $26,27$ using an Agilent 8719ET (0.05–13.5 GHz) network analyzer. The temperature coefficient of resonant frequency τ_f was also measured by the same method by changing temperature mainly from −30 to 80° C and calculated from the equation:

$$
\tau_{\rm f}(ppm/^{\circ}C) = \frac{f_{80} - f_{-30}}{f_{25} \times 110} \times 10^6
$$

where f_{80} , f_{-30} , and f_{25} are the resonant frequency at 80, -30 , and 25° C, respectively.

3. Results and discussion

Fig. 1 shows the XRD patterns of (Ca1−*x*,Mg*x*)SiO3 ceramics with the composition range of $0.1 \le x \le 0.5$. For the example of $x = 0.1$, the XRD profiles of the specimen can be indexed by the $CaSiO₃$ wollastonite structure. Compared with the theoretic XRD patterns of the CaSiO₃ wollastonite structure (JCPDS #84-0655), a slight shift in the position of maximum intensity peaks towards the higher side was observed, indicating a decrease in lattice parameter. The result is due to the substitution of Mg^{2+} with the lower ionic radius of 0.78 Å for Ca^{2+} with larger ionic radius of 1.06 Å in the A-site. The CaMgSi₂O₆ phase (JCPDS) #19-0239) with a C2/c monoclinic structure cell, $a = 9.748 \text{ Å}$, $b = 8.926 \text{ Å}$, $c = 5.250 \text{ Å}$, $\beta = 105.8^{\circ}$, and $Z = 4$ was observed at $x = 0.2$, which means that the maximum content of Mg solid solution in $CaSiO₃$ is lower than 0.2. The peak intensity of $CaMgSi₂O₆$ increased steadily and that of $CaSiO₃$ decreased with the increase of *x* in the range of 0.2–0.4. When $x = 0.5$, the phase $CaSiO₃$ disappeared and only the phase $CaMgSi₂O₆$ was observed.

Microwave dielectric properties of (Ca_{1−*x*},Mg_{*x*})SiO₃ ceramics sintered at $1300\,^{\circ}$ C for 3 h are presented in [Table 1.](#page-2-0) When the x value increased from 0.1 to 0.5, the dielectric constant ε_r increased nonlinearly from 6.49 to 7.46, the $Q \times f$ values firstly decreased from 62,420 to 48,784 GHz and then increased to 59,638 GHz, and no remarkable variations in the τ_f were

Fig. 1. XRD patterns of $(Ca_{1-x},Mg_x)SiO_3$ ceramics (0.1 ≤ *x* ≤ 0.5) sintered at 1300 °C for 3 h.

Table 1 Microwave dielectric properties of $(Ca_{1-x},Mg_x)SiO_3$ ceramics sintered at 1300 ◦C for 3 h

x value	$\varepsilon_{\rm r}$	$Q \times f$ (at 8 GHz)	τ_f (ppm/°C)
0.2	6.52	53,700	-43.2
0.3	6.79	48,800	-46.4
0.4	7.35	53,800	-45.7
0.5	7.46	59,700	-42.3

Table 2

Microwave dielectric properties of $(1 - y)$ CaMgSi₂O₆ + *y*CaTiO₃ ceramics sintered at 1300 ◦C for 3 h

<i>v</i> value	$\varepsilon_{\rm r}$	$Q \times f$ (at 8 GHz)	τ_f (ppm/ $^{\circ}$ C)
0.05	8.25	56,200	-27.5
0.08	8.78	54,100	-14.7
0.10	9.09	52,300	-6.9
0.12	9.42	50,800	$+5.6$
0.15	9.81	47,500	$+13.7$

observed. As shown in the XRD patterns [\(Fig. 1\),](#page-1-0) single-phased CaSiO₃, and CaMgSi₂O₆ were observed at composition $x = 0.1$ and 0.5, respectively. It indicated that the ε_r of CaSiO₃ and $CaMgSi₂O₆$ were 6.49 and 7.46, respectively. Thus, it could conclude that the ionic polarizability of $CaMgSi₂O₆$ is higher than that of CaSiO₃. In general, because a near-zero τ_f value of dielectric ceramics is necessary for commercial applications, an improvement in τ_f value, closer to 0 ppm/ \degree C, is required in this system.

In order to acquire the closer zero τ_f , CaTiO₃ was added to $CaMgSi₂O₆$. Table 2 demonstrates the microwave dielectric properties of $(1 - y)$ CaMgSi₂O₆–*y*CaTiO₃ ceramic system sintered at $1300\degree$ C for 3 h. As *x* value increased from 0.05 to 0.15, the τ_f values of $(1 - y)$ CaMgSi₂O₆-yCaTiO₃ ceramics varied from -27.5 to 13.7 ppm/ \degree C. Since the τ_f went through zero, it indicates that zero τ_f value can be obtained by appropriately adjusting the *y* value of $(1 - y)$ CaMgSi₂O₆–*y*CaTiO₃ ceramics. The XRD patterns of $0.88CaMgSi₂O₆ - 0.12CaTiO₃$ sintered at 1300 \degree C for 3 h is shown in Fig. 2(e). The XRD patterns showed

Fig. 2. XRD patterns of ceramics with (a) 0.5 wt\% , (b) 1 wt $\%$, (c) 1.5 wt $\%$, and (d) 2 wt% Li₂CO₃–V₂O₅ addition sintered at 880 °C for 2 h, and (e) no addition sintered at 1300 °C for 3 h.

that peaks indicating the presence $CaMgSi₂O₆$ as the main crystalline phase, in association with $CaTiO₃$ phase. It also indicated that solid solution did not form in $CaMgSi₂O₆ - CaTiO₃$ ceramic system due to the structure difference of $CaMgSi₂O₆$ (monoclinic structure) and $CaTiO₃$ (orthorhombic structure).

To further decrease the sintering temperature of this new microwave dielectric ceramic 0.88 CaMgSi₂O₆-0.12CaTiO₃, small contents of $Li_2CO_3-V_2O_5$ were doped into the samples. Fig. 2(a)–(d) presents the XRD patterns of the ceramics sintered at 880 °C for 2 h with different amounts of $Li_2CO_3-V_2O_5$. It could be observed that all the compounds exhibit the mixture of $CaMgSi₂O₆$ phase and $CaTiO₃$ phase, similar to those shown in Fig. 2(e). The SEM images of $0.88CaMgSiO₆-0.12CaTiO₃$ ceramics with 1 and 1.5 wt% Li₂CO₃-V₂O₅ sintered at 880 °C are shown in Fig. 3. Probably because of the liquid phase, the denser microstructures and almost no pores were found.

[Fig. 4](#page-3-0) shows the densities and microwave dielectric properties of the $0.88CaMgSi₂O₆ - 0.12CaTiO₃$ ceramics with $Li₂CO₃ - V₂O₅$ sintered at optimal temperatures. The optimum temperature for samples with 0.5 and 1–2 wt% additive were 930 and 880° C. From [Fig. 4, i](#page-3-0)t can be seen that the densities of sam-

Fig. 3. SEM micrographs of the polished surface of the ceramics with (a) 1 wt% Li₂CO₃–V₂O₅ and (b) 1.5 wt% Li₂CO₃–V₂O₅ sintered at 880 °C for 2 h.

Fig. 4. Densities and microwave dielectric properties of 0.88 CaMgSi₂O₆– 0.12CaTiO₃ ceramics with $Li_2CO_3-V_2O_5$ content of 0.5 and 1–2 wt%, sintered for 2 h at 930 and 880 ◦C, respectively.

ples with $0.5-2$ wt% were more than 3.12 g/cm³, which means that the $Li_2CO_3-V_2O_5$ is a very effective low-temperature additive. The density decreased slowly as the $Li₂CO₃-V₂O₅$ content increased. The dielectric constant ε_r changed in manner similar to density, which decreased slightly from about 9.35 to 9.06 as the $Li_2CO_3-V_2O_5$ content increased from 0.5 to 2 wt%. The decrease in $Q \times f$ value was probably due to the existence of unknown second phase with the increase of Li₂CO₃–V₂O₅. As increasing Li₂CO₃–V₂O₅ content, τ_f tends to be shifted to negative region linearly and varied from 3.9 ppm/ $\rm{°C}$ at 0.5 wt% Li₂CO₃–V₂O₅ addition to −4.4 ppm/ $\rm{°C}$ at 2 wt% Li₂CO₃-V₂O₅ addition. This implies that the τ_f values of the ceramics could be adjusted to near zero by adding slightly more $Li_2CO_3-V_2O_5$. Thus, it is obvious that these ceramics might be good candidates for LTCCs, and for 1 wt% Li₂CO₃–V₂O₅, ceramics with density $\rho = 3.22$ g/cm³, ε _r = 9.23, $Q \times f = 46,200$ GHz, and $\tau_f = 1.3$ ppm/ \degree C could be obtained at $T_s = 880 °C$.

In order to investigate ceramics slurry for tape casting, the powders were first mixed with solvents and dispersant in a ball mill for 24 h, and then plasticizers and binder were added, and mixed for another 24 h to obtain the slurry. The solvents were toluol and ethanol. The dispersant was menhaden fish oil (OA85). The binder itself was polyvinyl butyral (BM-2), and the compatible plasticizer was butyl benzyl phthalate (S160). The tape casting was done with caster (TM-MC-500). The ceramics slurry viscosity was 1200 mPa S. Generally, ceramics slurries with viscosity of 500–2500 mPa S are suitable for the tape cast-ing process.^{[28](#page-4-0)} Fig. 5 shows the microstructures of green tapes. In the green state, the average particle sizes of LTCC ceramics was around $1 \mu m$, and the microstructures of green tapes were uniform and there was no existence of agglomerates. On the other hand, the surface of green tape was glabrous; the green tape had very high density and tensile strength.

For compatibility tests, ceramics sheet with Ag electrodes were co-fired and analyzed to detect interactions between the low-fired samples and electrodes. SEM analysis revealed no interaction forming new phases after firing, as shown in Fig. 6. It is obvious that reaction of LTCC materials with Ag elec-

Fig. 5. SEM micrograph of the green tape.

Fig. 6. SEM micrograph of 0.88 CaMgSi₂O₆-0.12CaTiO₃ ceramics with 1 wt%. Li₂CO₃–V₂O₅ samples co-fired with Ag in air at 880 °C for 2 h.

trodes did not occur. Therefore, $CaMgSi₂O₆ - CaTiO₃$ with $Li_2CO_3-V_2O_5$ could be selected as suitable candidates for LTCC materials, because of low sintering temperature, good microwave dielectric properties, and compatibility with electrodes.

4. Conclusions

In this study, phase constitutes and dielectric properties of $(Ca_{1-x}, Mg_x)SiO_3$ $(x=0.1-0.5)$ ceramics were studied with different x contents, and CaTiO₃ demonstrated an effective compensation in $(Ca_{1-x},Mg_x)SiO \tau_f$ value. The $0.88CaMgSi₂O₆ - 0.12CaTiO₃$ composition was selected, due to its reasonable dielectric properties: $\varepsilon_r = 9.42$, $Q \times f = 50,800 \text{ GHz}$, and $\tau_f = 5.6 \text{ ppm} / \text{°C}$. The addition of ≤ 2.0 wt% Li₂CO₃-V₂O₅ was very effective in lowering the sintering temperature (T_s) , and dense ceramics could be obtained at $T_s \leq 900$ °C. The addition of Li₂CO₃–V₂O₅ does not induce apparent degradation in the microwave properties. It was found that dielectric constant (ε_r) and the quality factor $(Q \times f)$ decreased, the temperature coefficient of resonant frequency (τ_f) shifted to a negative value with the increase amount of $Li_2CO_3-V_2O_5$ addition. The specimens with 1 wt% Li₂CO₃-V₂O₅ sintered at 880 °C for 2 h shows excellent dielectric properties: $\varepsilon_r = 9.23$, $Q \times f = 46,200$ GHz, and τ_f = 1.3 ppm/ \degree C. The low-temperature sintering ceramics powders were suitable for the tape casting process. Also, the material is compatible with Ag electrodes and, therefore, is suitable for LTCC application.

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